

- 1) The name of the compound whose formula is Ca_3N_2 is
- calcium nitrate
 - nitrogen calsite
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- 2) What is formed when an atom loses electrons?
- covalent bond
 - a molecule
 - different element
 - positive ion
- 3) A type of chemical bond which often formes between two nonmetals is
- covalent
 - ionic
 - metallic
 - molecular bond
- 4) An element which has the same valence electron configuration as tin (Sn) is
- Si
 - Au
 - Ca
 - Xe
- 5) Which of the following is not an ionic compound?₂
- NaBr
 - CaH₂
 - H₂S
 - BaO
- 6) The atomic mass of the element is the number of:
- protons in its nucleus
 - neutrons in its nucleus
 - protons and neutrons in the nucleus
 - electrons in the shell

- 7) A neutral atom contains 5 electrons and 5 protons. This must be an atom of
- a) could be several atoms
 - b) boron
 - c) nitrogen
 - d) neon
- 8) what type of bond exists in a molecule of ammonia?
- a) ionic
 - b) covalent polar
 - c) covalent nonpolar
 - d) metallic bonds
- 9) When two atoms equally share two electrons this bond is
- a) polar and double
 - b) polar and single
 - c) nonpolar and single
 - d) ionic
- 10) The number of oxygen atoms in the correct formula of aluminium oxide is:
- a) 0
 - b) 1
 - c) 2
 - d) 3
- 11) A calcium ion
- a) has one electron in its outer energy level
 - b) has two electrons in its outer energy level
 - c) has a filled outer energy level
 - d) lacks two electrons of having noble gas configuration
- 12) The atoms with the same number of electron shells are in the same
- a) period
 - b) group
 - c) column
 - d) zone

- 13) How many sublevels are present in the third energy level?
- a) 2
 - b) 3
 - c) 4
 - d) 10
- 14) Which of the following designates the sublevels that exist in the energy level 2?
- a) s, p
 - b) p, f
 - c) s, p, d
 - d) s, p, d, f
- 15) The atom of which element has two unpaired electrons in the 2p sublevel?
- a) N
 - b) C
 - c) Be
 - d) F
- 16) Which of the following combinations represents an ion with charge +1 and mass number 85?
- a) 48 neutrons, 37 protons, 37 electrons
 - b) 48 neutrons, 37 protons, 36 electrons
 - c) 48 neutrons, 36 protons, 37 electrons
 - d) 48 neutrons, 35 protons, 36 electrons
- 17) An element with seven electrons in the outer level would be a ...
- a) metal
 - b) metalloid
 - c) nonmetal
 - d) nitrogen
- 18) In which series the elements have metallic character:
- a) Br, P
 - b) Ba, Cs
 - c) Al, S
 - d) B, Cs

19) Atoms of the chemical elements are characterized by the number of ...

- a) elemental forms
- b) electrons
- c) protons
- d) neutrons

20) When they combine with nonmetal atoms, metal atoms tend to

- a) gain electrons to become negative ions
- b) lose electrons to become positive ions
- c) remain electrically neutral with a charge of zero
- d) any of the above depending on the circumstances